



GUJARAT TECHNOLOGICAL UNIVERSITY

Bachelor of Engineering

Subject Code: 3163517

Semester – VI

Subject Name: Analytical Techniques

Type of course: Open Elective

Prerequisite: Needs basic knowledge of Chemistry

Rationale: The main objective of this subject is to provide a strong basis of Analytical chemistry that will be applicable to other areas of the degree course such as chemical reaction engineering. It also helps for assurance of quality, safety and efficacy of drugs, pharmaceuticals and of any compound.

Teaching and Examination Scheme:

Teaching Scheme			Credits	Examination Marks				Total Marks
L	T	P		Theory Marks		Practical Marks		
			ESE (E)	PA (M)	ESE (V)	PA (I)		
2	0	2	3	70	30	30	20	150

Content:

Sr. No.	Content	Total Hrs.
1	Fundamentals of Analytical Chemistry: Concept of quality: Definition of quality, Quality control & assurance, TQM. Correlation between quality & analysis, steps & types of chemical analysis, Stoichiometry & expression of concentration.	04
2	Theory of errors: Sources & classification of errors. Statistical treatment of analytical data & presentation of result. Sampling of solids, liquids & gases. Evaluation & validation of analytical methods. Good laboratory practices.	04
3	Chromatographic methods: Introduction & classification of chromatography. Theory, instrumentation & applications of the following chromatographic techniques: (i) Column chromatography (ii) TLC (iii) Paper chromatography (iv) GC (v) HPLC	15
4	UV-Visible Spectroscopy: Introduction, Theory of UV- Visible Spectroscopy & colourimetry, Beer Lambert law, Deviation from Beer Lambert law. Infrared Spectroscopy: Introduction, Infrared radiation & its interaction with organic molecules, vibrational mode of bonds, instrumentation & applications, interpretation of IR spectra. Nuclear magnetic resonance spectroscopy:	15



GUJARAT TECHNOLOGICAL UNIVERSITY

Suggested Specification table with Marks (Theory) Engineering

Subject Code: 3163517

Distribution of Theory Marks

R Level	U Level	A Level	N Level	E Level	C Level
60	13	9	9	9	0

Legends: R: Remembrance; U: Understanding; A: Application, N: Analyze and E: Evaluate C: Create and above Levels (Revised Bloom's Taxonomy)

Note: This specification table shall be treated as a general guideline for students and teachers. The actual distribution of marks in the question paper may vary slightly from above table.

Reference Books:

1. Instrumental Methods of Chemical Analysis, E. W. Ewing, McGraw Hill, New York. 4th Ed, 1975
2. Instrumental Methods of Analysis, B. K. Sharma, Goel Publishing house.
3. Elementary Organic Spectroscopy, Y.R. Sharma, S.Chand & company Ltd. New Delhi 2008

Course Outcomes: After learning this course students will be able to

Sr. No.	CO statement	Marks % weightage
CO-1	To express fundamentals of Analytical Techniques.	35
CO-2	To understand the working of instruments as well as for the development of new technologies.	35
CO-3	It provides assurance of quality, safety and efficacy of drugs, pharmaceuticals and of any compound.	30

List of Practicals:

1. Separation using Paper Chromatography
2. Separation using Thin Layer Chromatography
3. Volumetric Estimation of alloys
4. To find out concentration of unknown solution using Colourimeter
5. Separation using Column Chromatography
6. Working and principle of UV Spectrophotometer
7. Complexometric Titration by EDTA
8. Gravimetric Estimation
9. Potentiometric Titration of between BaCl₂ and K₂CrO₄
10. Turbidity meter



GUJARAT TECHNOLOGICAL UNIVERSITY

Bachelor of Engineering

Subject Code: 3163517