A Specially Designed Initiative to Encourage Young Talent by



TALLENTEX 2017 : (23, October 2016)

PAPER CODE







Duration: 2 Hrs. | Maximum Marks: 320

Tallentex Roll No. **Answer Sheet No.**

Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.

Things NOT ALLOWED in EXAM HALL: Blank Paper, clipboard, log table, slide rule, calculator, camera, mobile and any electronic or electrical gadget. If you are carrying any of these, then keep them at a place specified by invigilator at your own responsibility.

INSTRUCTIONS

- This Booklet is your Question Paper. DO NOT break seal of Booklet until the invigilator instructs to do so.
- Fill your TALLENTEX Roll No. & Answer Sheet No. in the space provided on the cover page.
- Carefully fill your PAPER CODE and present CLASS in space provided (Serial No. 6 & 12) of optical response sheet.
- 4. Please make sure that paper you received is of your class only.
- 5. The Answer Sheet is provided to you separately which is a machine readable Optical Response Sheet (ORS). You have to mark your answers in the ORS by darkening bubble, as peryour answer choice, by using black or blue ball point pen.
- 6. After breaking the Question Paper seal, check there are 16 pages in the booklet. This Question Paper contains 80 MCQs with 4 choices (Subjects: Mental ability: 1-20, Physics: 21-40, Chemistry: 41-60, Biology: 61-80 / Maths: 61-80) Important: Attempt Only One Subject from Biology / Mathematics.
- 7. Think wisely before darkening bubble as there is negative marking for wrong answer. Answer once marked by pen cannot be cancelled.
- 8. Marking Scheme:
 - a. If darkened bubble is RIGHT answer: 4 Marks.
 - b. If darkened bubble is WRONG answer: -1 Mark (Minus One Mark).
 - c. If no bubble is darkened in any question: No Mark.
- 9. If you are found involved in cheating or disturbing others, then your ORS will be cancelled.
- 10. Do not put any stain on ORS and hand it over back properly to the invigilator.
- 11. You can take along the question paper after the test is over.
 - * Fill appropriate circle of subject in column no. 12 of ORS, otherwise your ORS will be treated as invalid.

ALLEN RESULT: JEE ADVANCED-2016

4 in Top 10 | 12 in Top 50 | 25 in Top 100 AIR



AIR: 2



Bhavesh Dhingra Classroom



Kunal Goyal



Classroom



AIR: 9

Gaurav Didwania Classroom



Rohan Garg Classroom



Animesh Bohra Distance



Ritesh Goenka Classroom

AIR: 27



Vikrant Garg Classroom



Sharvik Mittal Classroom



Ishan Tarunesh Distance



Naman Jain Classroom



Sushil Khyalia Classroom

Total Selections

Classroom: 2857 | Distance: 1026

ALLEN RESULT: NEET (UG)-2016

7 in Top 10 | 35 in Top 50 | 58 in Top 100 AIR



AIR: 2



Ekansh Goyal Classroom



Nikhil Bajiya



Classroom



Ashank Khaitan Distance



Dyuti Shah Distance



Japnoor Kaur . Distance



Utkarsh Anand Classroom



Prakhar Bansal Classroom



Lajjaben Patel Classroom



Gurasis Singh Distance



Swetank Anand Classroom



Mahak Kr. Surana Classroom



Prachi Singh Classroom

Total Qualified

Classroom : 26198 | Distance : 6908

Authenticity of Result: Power of ALLEN

ALLEN RESULT: AIIMS-2016

8 in Top 10 | 25 in Top 36



AIR: 3



AIR: 4

Het Sanjay Shah Classroom



Mridul Sharma Classroom

AIR: 5



AIR: 6

Dyuti Shah Distance





AIR: 7

Aishvary Gupta Classroom



Kushagra Pandey Distance



Ekansh Goyal Classroom



Lajjaben Patel



Ira Pachori



Ritik M Goyal





Amol Sood Classroom





Dhruvil D. Shah





Classroom





Sanil Garo Distance



Aditya Agarwal Distance





Distance



Gurasis Singh Distance



Manavi Gupta Classroom



Prachi Singh Classroom



Japnoor Kaur Distance



Ayush Jain Classroom



Sukriti Chaudhri Distance

Total Qualified

Classroom : **405** | Distance : **197**

ALLEN RESULT: JEE Main-2016

8 in Top 100 | 25 in Top 200 | 65 in Top 500 | 136 in Top 1000



AIR-105

Classroom



Megh V. Thakkar Shashwat Agrawal Rohan Garg Amey Ravindra Patil Akash Bhardwaj Rahul Agrawal





Distance









Sharvik Mital Shashwat Shivam Ankit Dhankhar



Classroom





Sukriti Gupta Georgi Joseph Boby Distance

AIR-171

Rushikesh Vitthal Distance



Classroom



Koustav Yacha Rahul M. Chanduka Classroom

Students secured JEE Main



Gavali H. Abhiman Distance



Atri Dutta Distance



Vansh J. Chiripal Classroom

All India Ranks from all Courses of ALLEN

TALLENTEX Success Power Session & Rewards Ceremony (29 November 2015) Recognition & Reward at NATIONAL Level



SECTION-A: MENTAL ABILITY

This section contains **20 Multiple Choice Questions.** Each question has four choices (1), (2), (3) and (4) out of which ONLY ONE is correct.

1.	The following question	on is based on the follow	wing information:					
	(i) M % N means M i	s the son of N						
	(ii) M @ N means M	is the sister of N						
	(iii) M \$ N means M is the father of N							
	Which of the following	g shows the relation that	at C is the granddaughte	r of E?				
	(1) C % B \$ F \$ E		(2) B \$ F \$ E % C					
	(3) C @ B % F % E		(4) E % B \$ F \$ C					
2.	A told B, "Yesterday I	met the only brother of	the daughter of my grand	mother." Whom did A meet ?				
	(1) Cousin	(2) Brother	(3) Nephew	(4) Father				
3.	My office is to the Ea	st of my house. My siste	er's office is to the South	n-East of my office. A park is				
	to the South of my of	fice. In which direction	is my house located wit	th respect to the Park?				
	(1) North-East	(2) South-West	(3) North-West	(4) South-East				
4.	If $CACOPHONY = 8$	1, ALLEN= 25, PANOR	MA = 49, then TRANS	MOGRIFY =?				
	(1) 125	(2) 144	(3) 91	(4) 121				
5.	Today is Monday. Aft	ter 61 days, it will be						
	(1) Monday	(2) Wednesday	(3) Saturday	(4) Friday				
6.	Find the missing term	in the given series						
	235, 236, 234, 237, 2	233, 238, ?						
	(1) 232	(2) 235	(3) 237	(4) 231				
7.	A clock which gains u	iniformly is one minute	slow at Noon on 26/09/2	016. It is four minutes fast at				
	Noon on 30/09/2016.	When did it show the c	correct time ?					
	(1) 28/09/2016 at 7:13	2am	(2) 27/09/2016 at 7:0	6am				
	(3) 28/09/2016 at 7:0	6am	(4) 27/09/2016 at 7:13	2am				
8.	What is the angle bet	ween the hour and minu	ite's hand, when the clos	ck shows 30 minutes past 6?				
	(1) 30°	(2) 15°	(3) 10°	(4) 20°				
9.	A boy was born on 29	9th Feb, 1896. When wi	ill he celebrate his next	birthday ?				
	(1) 28 th Feb, 1897		(2) 29 th Feb, 1900					
	(3) 29 th Feb, 1904		(4) 20 th Feb, 1908					
10.	In the given figure a	solid cube is painted or	all sides by a single co	olor. Observe the given solid				
	and choose the correct	et alternative.						
		4						
			/// h					
			$\mathcal{A}(\mathcal{M})$					
		μΙΙ	 / <i> </i> /					

Find the total number of blocks whose only one surface is colored?

(1) 8

- (2) 12
- (3) 16
- (4) 24



11. Four standard dice are thrown on the ground. The total of numbers on the top faces of these four dice is 13 as the top faces showed 4, 3, 1 and 5 respectively. What is the total of the faces touching the ground?

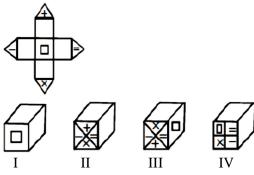
(1) 12

(2) 15

(3) 13

(4) 16

12. Select the correct answer choice(s) which is/are formed by folding the unfolded dice.



(1) I and II

(2) II and III

(3) I and III

(4) II and IV

13. Faces of a large cube are painted with six different colours-Red, Violet Yellow, Green, Orange and Blue. Green and Violet are opposite to each other. Red and Orange are opposite to each other. The cube is placed on a table with the Yellow face touching the table and the Orange face is towards the front. The cube is cut into 210 identical pieces by making the least number of cuts. Out of the total cuts made the maximum number of cuts are made in the horizontal direction and the least number of cuts in the direction parallel to the Violet face.

How many pieces have atmost one face painted?

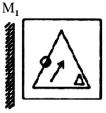
(1) 116

(2) 124

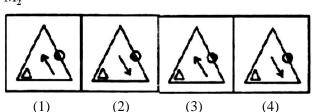
(3) 142

(4) 154

14. In the following question, choose the correct mirror image of the figure (X) from amongst four alternatives.



M,



15. There are six teachers A, B, C, D, E and F in a school. Each of the six teachers teaches two subjects, one compulsory subject and other optional subject. D's optional subject was History while three others have it as compulsory subject. E and F have Physics as one of their subjects. F's compulsory subject is Mathematics which is an optional subject of both C and E. History and English are A's subjects but in terms of compulsory and optional subject they are just reverse of those of D's. Chemistry is an optional subject of only one of them. The only female teacher in the school has English as her compulsory subject.

What is C's compulsory subject?

(1) History

- (2) Physics
- (3) Mathematics
- (4) English



ALLE	N'S Talent Encouragement Exam			02/100 /t/ / upor code				
16.		language 'they have	come back' is written	as 'na ja sa da' and 'they have g	one			
10.			is 'come' written in that	•	JIIC			
	(1) sa	(2) na	(3) ja	(4) sa or ja				
17.								
	Count the numbe	r of triangles and squ	ares in the figure					
	(1) 21 triangles,	7 squares	(2) 18 triangle	es, 8 squares				
	(3) 20 triangles, 8	8 squares	(4) 22 triangle	es, 7 squares				
18.	Give your answer	as:						
	(1) If only conclu	(1) If only conclusion I follows						
	(2) If only conclu	ision II follows						
	(3) If either I or I	II follows						
	(4) If neither I no	or II follows						
	Statement:	1. Some hats	are caps.					
		2. Some caps	are mats.					
	Conclusions:	I. Some caps	are hats.					
		II. Some mats	are hats.					
19.	Statements: Son	ne keys are staplers.	Some staplers are sticked	ers. All stickers are pens:				
	Conclusion:	(I) Some pens	are staplers					
		(II) Some stick	kers are keys					
		(III) No sticke	r is key					
		(IV) Some sta	plers are keys.					
	(1) Only (I) and	(II) follow						
	(2) Only (II) and	(III) follow						
	(3) Only (II) and	(IV) follow						
	(4) Only (I) and	(IV) and either (II) o	r (III)					
20.	If the first and th	e third letters in the	letter group DISTRIBU	TTION are interchanged and also	the			
	second and fourth	h letters, the fifth an	d seventh and so on th	en which of the following would	be			
	seventh letter from	m the left?						
	(1) U	(2) R	(3) B	(4) T				

SECTION-B: PHYSICS

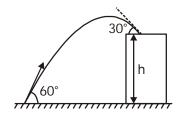
This section contains **20 Multiple Choice Questions**. Each question has four choices (1), (2), (3) and (4) out of which ONLY ONE is correct.

- 21. The postion vector of a particle is given as $\vec{r} = (t^2 4t + 6)\hat{i} + (t^2)\hat{j}$. The time, after which the velocity vector and acceleration vector becomes perpendicular to each other, is equal to—
 - (1) 1 sec
- (2) 2 sec
- (3) 1.5 sec
- (4) not possible



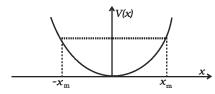
- An explosive of mass 18 kg is located at position vector $\vec{r}_0 = 6\hat{i} + 14\hat{j} + 20\hat{k}$ in gravity free space. 22. It breaks into four parts of masses 5kg, 3kg, 4kg and 6 kg respectively. After sometime the position vector of first three parts are $10\hat{i} + 20\hat{j} + 9\hat{k}, 8\hat{j} + 25\hat{k}$ and $4\hat{i} + 10\hat{j} + 20\hat{k}$ respectively. Find the position vector of the fourth part at that time?
 - $(1) \frac{\hat{\mathbf{i}} + 3\hat{\mathbf{j}} + \hat{\mathbf{k}}}{3}$
- (2) $3\hat{i} + \frac{29}{3}\hat{j} + \frac{80}{3}\hat{k}$ (3) $-\hat{i} + 2\hat{j} + 3\hat{k}$
- (4) None of these
- 23. Which of the following pair have same dimensions?
 - (1) Force and strain

- (2) Force and stress
- (3) Angular velocity and frequency
- (4) Energy and strain
- A stone projected at an angle of 60° from the ground level strikes at an angle of 30° on the roof 24. of a building of height 'h'. Then the speed of projection of the stone is-



- $(1) \sqrt{2gh}$
- $(2) \sqrt{6gh}$
- $(3) \sqrt{3gh}$
- $(4) \sqrt{gh}$
- 25. A block of mass M is moving on a smooth horizontal surface with constant speed u. Bullets are fired horizontally against the block to reduce the velocity of the block to half its initial value. Bullets get embedded in the block. Mass of each bullet is 'm' and speed 'u'. Then the number of bullets required is-
 - $(1) \frac{M}{3m}$
- $(2) \frac{M}{m}$
- $(3) \frac{M}{2m}$
- (4) None of these
- A bucket tied to a string is lowered at a constant acceleration of g/4. If the mass of the bucket is **26.** M and is lowered by distance d, the work done by the string will be (assume the string to be massless)
 - $(1) \frac{1}{4} Mgd$

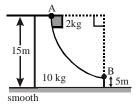
- (2) $\frac{-3}{4}$ Mgd (3) $\frac{-4}{3}$ Mgd (4) $\frac{4}{3}$ Mgd
- The potential energy function for a particle executing linear SHM is given by $V(x) = \frac{1}{2}kx^2$ where k 27. is the force constant of the oscillator (Fig.). For k = 0.5N/m, the graph of V(x) versus x is shown in the figure. A particle of total energy E turns back when it reaches $x = \pm x_m$. If V and K indicate the P.E. and K.E., respectively of the particle at $x = +x_m$, then which of the following is correct?



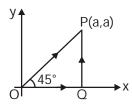
- (1) V = 0, K = E
- (2) V = E, K = 0
- (3) V < E, K = 0
- (4) V = 0, K < E.



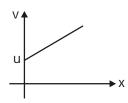
28. A block of mass 2 kg is released at point A on a rough platform ($\mu = 0.1$) of mass 10 kg as shown in figure. Block reaches point B with velocity 12 m/s w.r.t. platform find work done by friction?



- (1) 100 J
- (2) 80J
- (3) 56J
- (4) 100J
- A particle is moved from (0, 0) to (a, a) under a force $F = (3\hat{i} + 4\hat{j})$ from two paths. Path 1 is OP 29. and and path 2 is OQP. Let W₁ and W₂ be the work done by this force in these two paths. Then-

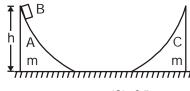


- (1) $W_1 = W_2$
- (2) $W_1 = 2W_2$
- (3) $W_2 = 2W_2$
- $(4) W_2 = 4W_1$
- **30.** A particle moves along x-axis in positive direction. Its acceleration 'a' is given as a = cx + d, where x denotes the x-coordinate of particle, c and d are positive constants. Velocity position graph of particle is as shown in following figure. The value of speed of particle at x = 0 should be—



- (1) $\sqrt{\frac{4d^2}{a^2}}$

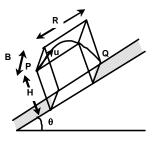
- Three forces are acting on a particle of mass m initially in equilibrium. If the first two force 31. (R₁ and R₂) are perpendicular to each other and suddenly the third force (R₃) is removed, then the acceleration of the particle is-
 - $(1) \frac{R_3}{m}$
- (2) $\frac{R_1 + R_2}{m}$ (3) $\frac{R_1 R_2}{m}$ (4) $\frac{R_1}{m}$
- In the given figure shown a small block B of mass m is released from the top of a smooth **32.** movable wedge A of the same mass m. The height of wedge A shown in figure is h = 100 cm. B ascends movable smooth wedge C of the same mass. Neglecting friction any where find the maximum height (in cm) attained by block B on wedge C



- (1) 25
- (2) 30
- (3) 35
- (4) 40

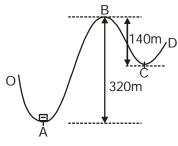


33. A box of dimension $H \times B \times R$ is kept on a frictionless inclined plane as shown in the figure. A stone is thrown from one corner P with speed u parallel to the ceiling of the box such that it hits the floor of the box. At the same instant, the box is released to slide down the plane. Calculate the time of flight.



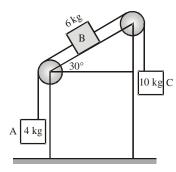
(3) $\sqrt{\frac{H}{g\cos\theta}}$

- (4) Cannot be calculated.
- 34. Track OABCD (as shown in the following figure) is smooth. What minimum speed has to be given to a particle lying at point A, so that is can reach point C? (take $g = 10 \text{ m/s}^2$)



- (1) 60 m/s
- (2) 100 m/s
- (3) 70 m/s
- (4) 80 m/s
- **35.** A spherical body of mass m and radius r is allowed to fall in a medium of viscosity η . The time in which the velocity of the body increases from zero to 0.63 times the terminal velocity (v) is called time constant (τ). Dimensionally, τ can be represented by-
 - (1) $\frac{\text{mr}^2}{6\pi n}$
- (2) $\sqrt{\frac{6\pi mr\eta}{g^2}}$ (3) $\frac{m}{6\pi \eta r \nu}$
- (4) None of these
- Three blocks A, B and C of mass 4 kg, 6kg and 10 kg respectively are connected as shown in **36.** figure. Find acceleration of block A?

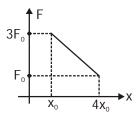
$$[g = 10 \text{ m/s}^2]$$



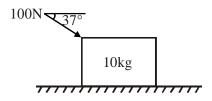
- (1) 10 m/s^2
- (2) 1.5 m/s² down
- (3) 3 m/s² upward
- (4) 1.5 m/s² upward



A particle of mass m moving along a straight line experience force F which varies with the distance traveled as shown. If the velocity of the particle at x_0 is $\sqrt{\frac{2F_0x_0}{m}}$, then velocity at $4x_0$ is—

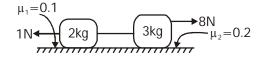


- (2) $2\sqrt{\frac{F_0 x_0}{m}}$
- (3) $\sqrt{\frac{F_0 x_0}{2m}}$
- In the figure shown below, calculate the angle of friction. The block does not slide. Take $g = 10 \text{ m/s}^2$.



- (1) $\tan^{-1}\frac{1}{2}$ (2) $\tan^{-1}\frac{1}{4}$ (3) $\cot^{-1}\frac{1}{2}$ (4) $\cot^{-1}\frac{1}{4}$
- The potential energy of particle of mass m free to move along x-axis is given by $U = \frac{1}{2}kx^2$ for 39. x < 0 and U = 0 for $x \ge 0$ (x denotes the x-coordinate of the particle and k is a positive constant). If the total mechnical energy of the particle is E, then its speed at $x = \sqrt{\frac{2E}{k}}$ is—
 - (1) zero

- $(2) \sqrt{\frac{2E}{m}} \qquad (3) \sqrt{\frac{E}{m}} \qquad (4) \sqrt{\frac{E}{2m}}$
- 40. In the following arrangement if f₁, f₂ and T be the frictional force on 2kg block, 3kg block and tension in the string respectively, then their values are-



- (1) 2N, 6N, 3.2 N
- (2) 2N, 6N, 0 N
- (3) 1N, 6N, 2 N
- (4) Data is insufficient to calculate the required values

41.

SECTION-C: CHEMISTRY

This section contains **20 Multiple Choice Questions.** Each question has four choices (1), (2), (3) and (4) out of which ONLY ONE is correct.

The wavelength emitted during the transition of electron in betweeen two levels of Li+2 ion whose

	sum is 4 and differen	nce is 2 is:			
	(1) 1.14×10^{-6} cm		(2) 11.4×10^{-6} c	m	
	(3) 0.114×10^{-6} cm		(4) 11.40×10^{-6}	cm	
42.	The NH ₃ evolved from	om 1.40 gm sampl	e of protein was absorbe	ed in 45 ml of 0.4 N HNO ₃ .	The
	excess acid required	20 ml of 0.1 M Na	aOH. The % N in the san	nple is	
	(1) 8	(2) 16	(3) 19.42	(4) None	
43.	λ_0 is the threshold w	avelength of a met	al for photoelectron emiss	sion. If the metal is exposed to	the
				21-	
	light of wavelength 7	then the maximu	m velocity of ejected elec	etron will be $\sqrt{\frac{2h}{m}}(\lambda_0 - \lambda) \times K$.	The
	value of K is:				
			(3) $\sqrt{\frac{c}{\lambda_0 \lambda}}$	$\sqrt{1}$	
	(1) speed of light	(2) 1	(3) $\sqrt{\lambda_0 \lambda}$	$(4) \sqrt{\frac{1}{\lambda \lambda_0}}$	
44.	1M HCl and 2M HC	Cl are mixed in the	e volume ratio of 4 : 1.	What is the final molarity of	HCl
	solution				
	(1) 1.5	(2) 1	(3) 1.2	(4) 1.8	
45.	The CORRECT ord	er of I.E ₂ is:			
	(1) $Ne > F > O > N$		(2) $O > F > Ne >$	> N	
	(3) Ne $> O > F > N$		(4) $O > Ne > F >$	> N	
46.	Three closed vessels	A, B and C are at	the same temperature T	and contain gases which obey	the the
	Maxwellian distribut	ion of velocities.	Vessel A contains only O	N_2 , B only N_2 and C a mixture	re of
	equal quantities of O	$\frac{1}{2}$ and N_2 . If the aver	rage speed of the O ₂ molecular	ales in vessel A is V_1 , that of the	ie N ₂
	molecules in vessel B	is V_2 , the average spe	eed of the O ₂ molecules in	vessel C is -	
	$(1) \ \frac{(V_1 + V_2)}{2}$		(2) V ₁		
	2		(-) 1		
	(3) $(V_1 \cdot V_2)^{1/2}$		$(4) \sqrt{3kT/M}$		
47.	How many milliliter	s of 0.1 M H_2SO_4	must be added to 50 ml	of 0.1 M NaOH to give a solu	ıtion
	that has a concentrat	tion of 0.05 M in I	H ₂ SO ₄ ? Assume no NaH	SO ₄ formation.	
	(1) 400 ml		(2) 200 ml		
	(3) 100 ml		(4) 50 ml		



48.	A 4:1 mixture of helium and methane is contained in a vessel at 10 bar pressure. Due to a hole in							
	the vessel, the ga	as mixture leaks out.	The molar composition of r	nixture effusing out initially is				
	(He: CH_4)							
	(1) 8 : 1	(2) 8 : 3	(3) 4:1	(4) 1 : 1				
49.	Find the group nu	imber and the number	of unpaired electrons present	in an element having following				
	atomic electronic	configuration of outerme	ost and penultimate shell (n-1)	$s^2p^6d^{10} ns^1$.				
	(1) 1,0	(2) 11,1	(3) 1,1	(4) 11,0				
50.	Consider 2 nd shell	l in a hypothetical sys	stem of atom given :-					
		n =	2 n = 2					
			(2p)					
	It is possible who	en value of magnetic o	quantum number are :-					
	(1) $-l$ to zero to $+l$ in integral steps							
	(2) $-(l-1)$ to zero to $+(l-1)$ in integral steps							
	(3) $-(l + 1)$ to zero to $+(l + 1)$ in integral steps							
	(4) - (l + 1) to ze	ro to $+(l-1)$ in integr	ral steps					
51.	Select the CORR	RECT order of ionic ra	adius:					
	(1) $H^+ > H$	(2) $H > H^-$	(3) $H^+ < H > H^-$	(4) $H^+ < H < H^-$				
52.	Maximum numbe	r of electrons that can	be accommodated in the M	th shell is:				
	(1) 2	(2) 8	(3) 18	(4) 32				
53.	The bond strengtl	n in $\mathrm{O_2}^+$, $\mathrm{O_2}$, $\mathrm{O_2}^-$ &	O ₂ ²⁻ follows the order:					
	$(1) O_2^{2-} > O_2^{-} > 0$	$O_2 > O_2^{+}$	$(2) O_2^+ > O_2^- > O_2^-$	$> O_2^{2-}$				
	(3) $O_2 > O_2^- > O_2$	$_{2}^{2-} > O_{2}^{+}$	$(4) O_2^{-} > O_2^{2-} > O_2^{-}$	$(4) O_2^- > O_2^{2-} > O_2^+ > O_2$				
54.	With the increase	in temperature of a ga	s, the fraction of molecules	having velocities within a given				
	range around the	most probable velocit	y (for same change in speed	d), would				
	(1) increase		(2) decrease	(2) decrease				
	(3) remain uncha	inged	(4) initially increase	e and then decrease				
55.	The INCORREC	T order of bond angle	e is:					
	(1) $CO_2 > CO_3^{2-1}$	> CF ₂ Cl ₂	(2) $NO_2^+ > NO_3^- >$	NO ₂				
	$(3) XeF_2 > XeF_4$	> XeO ₄	(4) $PH_3 > AsH_3 > S$	(4) PH3 > AsH3 > SbH3				
56.	Hybridisation of	'B' in the anionic part	of Borax is:					
	(1) sp only		$(2) sp^2 only$					
	(3) sp ³ only		(4) both $sp^2 \& sp^3$	(4) both sp2 & sp3				

57. The root mean square speed of the molecules of diatomic gas is u. When the temperature is doubled, the molecules dissociates into two atoms. The new rms speed of the atoms is :

(1) $\sqrt{2}u$

(2) u

(3) 2u

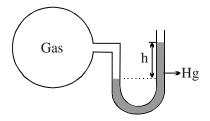
(4) 4u



- **58.** The **CORRECT** order regarding the electronegativity by hybrid orbitals of carbon is
 - (1) $sp < sp^3 < sp^2$

- (2) $sp < sp^2 < sp^3$ (3) $sp^2 > sp > sp^3$ (4) $sp > sp^2 > sp^3$
- A bulb of constant volume is attached to a very thin manometer tube as shown in figure. Gas starts **59.**

leaking through a small hole in the bulb causing change in pressure as $\frac{dP}{dt} = -kP^2$



Atmospheric pressure = 76 cm of Hg

Where \mathbf{k} is constant and \mathbf{P} is pressure at any instant.

Initial height difference 'h' was 76 cm and after 10 min 'h' was 38cm.

Calculate the value of \mathbf{k} in unit of $\mathbf{atm}^{-1}\mathbf{Hr}^{-1}$.

- (1) 1.00
- (2) 2.00
- (3) 0.5
- (4) 5.00
- 60. On reduction with hydrogen, 3.6 g of an oxide of metal left 3.2 g of metal. If the vapour density of metal is 32, the simplest formula of the oxide would be:
 - (1) MO
- (2) M_2O_3
- (3) $M_{2}O$
- $(4) M_{2}O_{5}$

Attempt any one of the Section-D (Biology) OR Section-E (Mathematics)

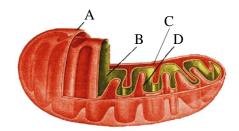
SECTION-D: BIOLOGY

This section contains 20 Multiple Choice Questions. Each question has four choices (1), (2), (3) and (4) out of which ONLY ONE is correct.

- Which of the following is correct regarding cells which are responsible for the hardness of fruit walls 61. of nuts?
 - (1) These cells are rich in protoplasm and are also found in the hypodermis of dicot stem
 - (2) These cells are loosely packed with large intercellular spaces and also perform various functions like photosynthesis, secretion etc.
 - (3) These cells are dead, without protoplast and also found in the leaves of tea
 - (4) These are continuously dividing cells and are responsible for producing the secondary tissues
- Which of the following statements regarding biological nomenclature is not correct? **62.**
 - (1) Biological names are generally in Latin
 - (2) The first word in a biological name represents the genus while the second component denotes the specific epithet
 - (3) The name of genus starts with a small letter while the specific epithet starts with a capital letter
 - (4) Both the words in a biological name are printed in italics to indicate their Latin origin



63.



To increase ATP production in above given cell organelle, the number or amount or diameter of which part among A, B, C or D should primarly increase?

(1) D

(2) C

(3) B

- (4) A
- **64.** Read the following features and identify the animal from the given options.
 - (a) Exoskeleton of chitin.
 - (b) Malphigian tubule as excretory organ.
 - (c) Tracheal system for respiration.
 - (d) 3 pair of legs in thoracic region.
 - (1) Limulus

(2) Prawn

(3) Spider

(4) Cockroach

65.



Which of the following option is incorrect about diagrammatic view of structure shown in the above figure?

- (1) It is cnidoblast or cnidocyte.
- (2) It contains the stinging capsule nematocyst.
- (3) It is found in Ctenoplana and used for anchorage, defence and for capture of prey.
- (4) It is found on the tentacles and the body of Cnidarians.
- 66. Conjoint, collateral, open and endarch vascular bundles are found in :-
 - (1) Dicot roots

(2) Monocot roots

(3) Dicot stems

- (4) Monocot stems
- 67. Key is another taxonomical aid used for _____ A ____ of plants and animals. The keys are based on the ____ B ____ characters generally in a pair called couplet. Keys are generally ____ C ___ in nature.

Choose the correct words respectively for A,B & C from the following options :-

- (1) Identification, Contrasting, Quantitative
- (2) Classification, Similar, Qualitative
- (3) Nomenclature, Common, Analytical
- (4) Identification, Contrasting, Analytical



68.	Which tissue provides	support and protection	to softer organs and tissues :-
	(1) Areolar connective	tissue	
	(2) Adipose connectiv	e tissue	
	(3) Bone		
	(4) Blood		
69.	Mark the option with	incorrect combination o	f the animal phylum and its unique characteristics ?
	(1) Porifera - Water va	ascular system which he	elps in locomotion
	(2) Mollusca - Raspin	g organ for feeding call	ed "radula"
	(3) Aschelminthes - P	seudocoelom developed	from embryonic blastocoel
	(4) Ctenophora - Cilia	ry comb plates for swir	nming
70.	Which phase is marke	d by terminalisation of	chiasmata?
	(1) Diakinesis		(2) Diplotene
	(3) Pachytene		(4) Zygotene
71.	Sclerenchyma: Lignin	n : : Collenchyma :	A
	Trichomes : Multicellu	ılar : : Root hairs :	B
	Sapwood: Water cond	duction: Heart wood	:C
	Choose the correct opt	ion for A,B and C from	the following options: -
	(1) $A = Suberin$	B = Unicellular	C = Food conduction
	(2) $A = Pectin$	B = Multicellular	C = Water conduction
	(3) $A = Lignin$	B = Multicellular	C = Food storage
	(4) A = Hemicellulose	B = Unicellular	C = Mechanical support
72.	If the microtubular arr	angement of axoneme o	f cilium or flagellum is replaced by the microtubular
	arrangement of centrio	le then after replacemen	nt, in cilium or flagellum :-
	(1) There will be 9 les	ss peripheral microtubul	es & 2 more central microtubules
	(2) There will be 2 les	ss peripheral microtubul	es & 9 more central microtubules
	(3) There will be 2 mg	ore peripheral microtubu	lles & 9 less central microtubules
	(4) There will be 9 mg	ore peripheral microtubu	lles & 2 less central microtubules
73.	Fluidity of plasma me	mbrane is due to-	
	(1) Glycoproteins		
	(2) Carbohydrates		
	(3) Lipids		
	(4) Proteins		



74. The four sketches (A, B, C and D) given below, represent four different types of animal tissues. Which one of these is correctly identified in the options given, along with its correct location and function?







(C)

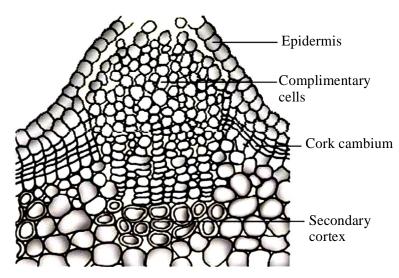


(D)



	Tissue	Location	Function
1	(C) Collagen fibres	Cartilage	Attach skeletal muscles to bones
2	(D) Smooth muscle tissue	Heart	Heart contraction
3	(A) Columnar epithelium	Nephron	Secretion and absorption
4	(B) Glandular epithelium	Intestine	Secretion

75.



Identify the above given figure and choose the correct option from the following:-

- (1) Stomata
- (2) Axillary bud
- (3) Lenticel
- (4) Leaf Primordium

- 76. Ureter and Urinary bladder are internally lined by -
 - (1) Stratified squamous epithelium
 - (2) Stratified columnar epithelium
 - (3) Transitional epithelium
 - (4) Stratified cuboidal epihelium
- 77. If in a cell 10 tetrads are present in prophase-I, then what will be the number of chromatids in each cell during Metaphase I, Metaphase-II, Anaphase I and Anaphase-II?

	Metaphase - I	Metaphase - II	Anaphase - I	Anaphase - II
(1)	40	20	40	20
(2)	40	10	20	10
(3)	20	40	20	40
(4)	10	20	10	20



(4) 7

78.	Which one is a not an example of Urochordata?									
	(1) Ascidia	(2) Salpa	(3) Doliolum	(4) Amphioxus						
79.	In animal cells, lipid	like steroidal hormone	es are synthesised in –							
	(1) RER	(2) SER	(3) Golgi complex	(4) Mitochondria						
80.	Which of the follow	ing phylum includes tr	iploblastic & metameric	cally segmented animals having						
	closed circulatory system :-									
	(1) Aschelminthes	(2) Annelida	(3) Mollusca	(4) Hemichordata						
		SECTION-E:	MATHEMATICS	S						
This	section contains 20 M	Iultiple Choice Question	ons. Each question has	four choices (1), (2), (3) and (4)						
out o	of which ONLY ONE	is correct.								
61.	$\sum_{n=1}^{\infty} \frac{n}{4n^4 + 1} \text{ equals to}$									
	(1) 0	(2) 1	(3) $\frac{1}{2}$	$(4) \frac{1}{4}$						
62.	If roots of the equation	on $x^2 + x + 1 = 0$ are	α and β then $(1-\alpha)$ $(1-\beta)$	B) is equal to						
	(1) 1	(2) 2	(3) 3	(4) 4						
63.	The least integral val	tue of k such that $(k-2)$	$(2)x^{2} + 8x + k + 4$ is posit	ive for all real values of x is						
	(1) 1	(2) 2	(3) 3	(4) 5						
64.	If z is a complex n	umber satisfying the eq	uation $ z - (1 + i) ^2 =$	$= 2$ and $\omega = \frac{2}{z}$,						
	then the locus traced	by 'ω' in the complex	plane is							
	(1) x - y - 1 = 0	(2) x + y - 1 = 0	(3) x - y + 1 = 0	(4) x + y + 1 = 0						
65.	Sum of maximum an	nd minimum value of the	ne function $f(x) = \sin^2 x$	$+ 8\cos x - 7 \text{ is}$						
	(1) –4	(2) -5	(3) 4	(4) -14						
66.				. units then k is equal to:						
	(1) 12	(2) 24	(3) 48	(4) 36						
67.	$\frac{x}{a} + \frac{y}{b} = 1$ touches the	circle $x^2 + y^2 = r^2$ then	the point $\left(\frac{1}{a}, \frac{1}{b}\right)$ lies	on						
	$(1) x^2 + y^2 = 1$	(2) $x^2 + y^2 = \frac{1}{r^2}$	(3) $x^2 + y^2 = r^2$	$(4) x^2 + y^2 = 2r^2$						
68.	If $ x^2 - 4 + x^2 - 9 $	= 5 for								
	$x \in [\alpha, \beta] \cup [\gamma, \delta]$ then	$ \alpha + \beta + \gamma + \delta $ is	equal to							
	(1) 20	(2) 6	(3) 10	(4) 12						
69.	Let α , β , γ , δ be the	e roots of $x^4 - x^3 - x^2 - 3$	1 = 0. Also consider $p(x)$	$= x^6 - x^5 - x^3 - x^2 - x$. Then the						
	value of $p(\alpha) + p(\beta)$	value of $p(\alpha) + p(\beta) + p(\gamma) + p(\delta)$ is equal to								

(3) 6

(1) 4

(2) 5



70.	$\sum_{n=1}^{100} (-1)^n .n is equal to$							
	(1) 25	(2) 50	(3) 75	(4) 100				
71.		through the points A(3 s k times AB, then value		th the x-axis. If the distance				
	(1) $20\sqrt{2}$	<u>_</u>	(3) $2\sqrt{10}$	(4) 6				
72.	If lines $(\lambda \alpha)x + y + 7$	$= 0, \ \mu x + (\lambda \alpha)y + 3 = 0$	0 and $-x + y + 4 = 0$ are	e concurrent where $\alpha \in R$ and				
	$\lambda \neq 0$ then greatest po	ossible integral value of	μ is					
	(1) 1	(2) 2	(3) 3	(4) 4				
73.	Number of integral val	lue(s) of x satisfying the	equalities $2x + 3 \le 6x$	-1 and $\frac{x+3}{x-2} \ge 4$, is				
	(1) 0	(2) 1	(3) 2	(4) 3				
74.	Let α and β be the root	ots of the quadratic equa	ation $x^2 - 2x - p = 0$. If	the line $\alpha x + \beta y = 4$ always				
			e of $a + b$ is (p,α,β,a,b)					
	(1) 4	(2) –4	(3) 0	(4) -1				
75.	The sum of the first three terms of an increasing G.P. is 21 and the sum of their squares is 189. Then the sum of its first n terms is							
	$(1) \ 3 \ (2^n - 1)$	(2) $12\left(1-\frac{1}{2^{n}}\right)$	(3) $6\left(1-\frac{1}{2^{n}}\right)$	$(4) \ 6 \ (2^n - 1)$				
76.	If Z be any complex n	number such that $ 4Z - 1 $	+ 4Z + 3 = 4 then lo	cus of Z is				
	(1) A circle	(2) An ellipse		(4) none of these				
77.			ent to the circles $(x + 2)^2$ 0. Then $b^2 - c$ is equal to	$y^2 + y^2 = 1$ and $(x - 4)^2 + y^2 = 9$				
	(1) 39	(2) 93	(3) 36	(4) 35				
78.	$\tan (\theta - \alpha) = a$ and $\tan \alpha$	$(\theta + \alpha) = b$ then tan 20	α equals to					
	$(1) \ \frac{a+b}{1-ab}$	$(2) \frac{b-a}{1+ab}$	$(3) \ \frac{a-b}{1+ab}$	(4) none of these				
79.	If the fourth roots of u	unity are z_1, z_2, z_3, z_4 , th	en $z_1^2 + z_2^2 + z_3^2 + z_4^2$ is eq	ual to				
80.	(1) 1	(2) 0	(3) i	(4) none of these y + 1 = 0 which are of unit				
	$(1) (x - 1)^2 + (y - 1)^2$	$=\frac{3}{4}$	(2) $(x-1)^2 + (y-1)^2$	= 2				
	$(3) (x - 1)^2 + (y - 1)^2$	= 4	(4) none of these					



SPACE FOR ROUGH WORK

ALLEN System



Orientation Session



Classroom Session



Prarthana



Open Session & Medal Distributions



Regular Test



Test Result - (CSAT)



Continuous Communication



Doubt Removal Counters



Online Practice Lab

Comprehensive Study Material	Ultimate Care	Board Work Sheets, Booklets
RACE : Regular Analysis through Continuous Exercise	Best Faculties	Expert Counselling



ALLEN Students Bring Glory to Nation through their International Achievements-2016

International Chemistry Olympiad



48th International Chemistry Olympiad IChO-2016 TBILISI, GEORGIA



SHARVIK MITTAL

International Physics Olympiad



47th International Physics Olympiad IPh0-2016 SWITZERLAND



DIVYANSH GARG (Classroom)

International Biology Olympiad



27th International Biology Olympiad IBO-2016 HANOI, VIETNAM



Gold Medal

LAJJA BEN PATEL

(Classroom)



Silver Medal
VIDUSHI VARSHNEY

International Earth Science Olympiad



10th International Earth Science Olympiad (IESO) 2016 JAPAN



Silver Medal

AMARJIIT VIKAS PANDE

(Classroom)

ALLEN Results: Pre-Nurture & Career Foundation (2015-16)

STAGES OF OFFICIAL OLYMPIADS MENTORED BY HBCSE

IJSO

STAGE 1

35 Selections in NSEJS STAGE 2

16 Selections in INJSO STAGE 3

3 Selections in OCSC NISHANT ABHANGI AYUSHMAN TRIPATHY GAURANG

selected for IJSO 2016

International Junior Science Olympiad



IB0

STAGE 1

3 Selections in NSEB STAGE 2

3 Selections in INBO

3 Selections in OCSC VIDUSHI VARSHNEY Got Silver Medal In

International Biology Olympiad

JEEVESH is the youngest in the country so far to qualify stage-1 of IBO

IChO

STAGE 1

1 Selection in NSEC DHYEY SANKALP GANDHI

is the youngest in the country so far to qualify stage-1 of IChO International Chemistry Olympiad



IOAA

STAGE 2 RAYYAN SHAHID selected in INJAO 2016

International Astronomy Olympiad Junior



IMO

STAGE 1

IBO 2016

190 Selections from Gujarat & 11 fro

from Gujarat & 11 from Maharashtra for RMO through PRE RMO.

3 STUDENTS SECURED 100% MARKS STAGE 2

14 Selections in RMO for INMO

International Maths Olympiad



IESO

STAGE 1

NET - 6 Selections Conducted by Geological Society of India International Earth Science Olympiad



IJSO

IJSO-2015



12th International Junior Science Olympiad (IJSO) 2015 KOREA



Gold Medal BHASKAR GUPTA (Classroom)



Gold Medal LAKSHYA SHARMA (Classroom)



Silver Medal VIDUSHI VARSHNEY (Classroom)

International Junior Science Olympiad

APTITUDE IN SCIENCE / MATHEMATICS

NSO

571 Selections in NSO (Level-1)

NISHANT ABHANGI: AIR-1 (Level-2)

NSO **National** Science **Olympiad**

Conducted by Science Olympiad Foundation, New Delhi



STSE 2015

36 Selections for Scholar Certificate

155 Selections for **Distinction Certificate**

Conducted by Rajasthan Board of Secondary Education

STSE State Talent Search **Examination**



NSTSE

232 Selections in NSTSE (Level-1)

63 Selections in NSTSE (Level-2)

NISHANT ABHANGI: AIR-1 (Level-2)

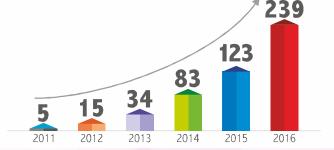
NSTSE National Science Talent Search **Examination**

Conducted by Unified Council, Hyderabad Test Assess Achieve

NTSE

Students Selected

NTSE 2016 (STAGE-2)



APTITUDE IN MATHEMATICS

NMTC

274 Selections in NMTC (Prelim)

17 Selections in NMTC (Final)

Conducted by Association of Mathematics Teachers of India, Chennai



NMTC



IMO (SOF)

370 Selections in IMO (Level-1)

IMO International **Mathematics Olympiad**

Conducted by Science Olympiad Foundation, New Delhi



APTITUDE IN <u>INFORMATION TECHNOLOGY</u>

140 Selections in UCO (Level - 1)

59 Selections in UCO (Level - 2)

UCO Unified Cyber Olympiad

Conducted by Unified Council, Hyderabad

SCIENTIFIC APTITUDE

BALSHREE HONOUR

14 Selections in **Balshree in Local Round** Rajasthan-7 | Gujrat-5 Madhya Pradesh-2)

Consist of Plaque, Citation, ₹15000 & Literature set



Conducted by National Bal

LANGUAGE PROFICIENCY

TRINITY GESE

38 Selections in **TRINITY GESE** Distinction: 21 Merit: 17

GESE Grade **Examination** for Spoken English

Trinity College, London



WORKSHOP/CONFERENCES

NMC

9 Selections in NMC **Including Ranks** 1,2,3 & 4

NMC National Maths Conference

Conducted by Association of Mathematics Teachers of India, Chennai



APTITUDE IN INTELLIGENCE QUATIENT (IQ)

TECHNOTHLON PRELIMS 2015

20 Students (10 Teams) **Selected for Techniche**

29 Selections for Silver Certificate in Technothlon Prelims

> Conducted by IIT Guwahati Techn®thlon

TECHNICHE 2015

2 Students (1 Team) **KHUSHI TIBAREWAL STUTI SHAH**

won Junior Squad in Techniche

Conducted by IIT Guwahati

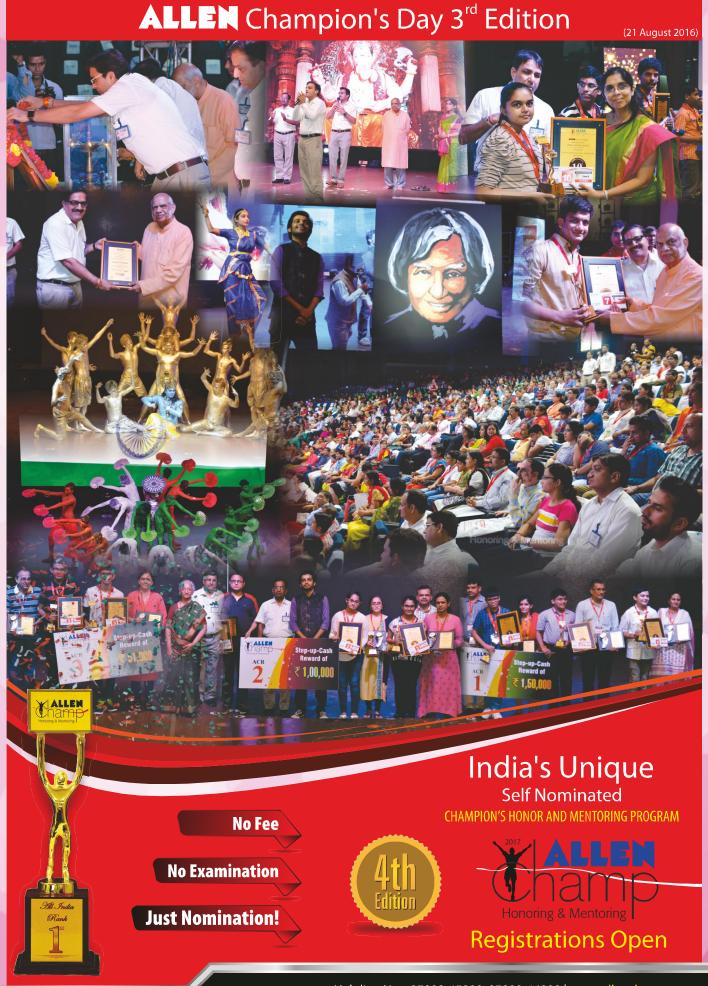


TECHKRITI

7 Students (Including AIR-1 & AIR-3) Selected in Techkriti

Conducted by IIT Kanpur





Answer Key



Class- 11th (XI)

Held on: 23 October 2016

M	ent	al Ability		Pł	nysics		Che	mistry		Bi	ology	IV	lath	ematics
Q. No.	Ans.		Q. No.	Ans.		Q. No.	Ans.		Q. No.	Ans.		Q. No.	Ans.	-
1	3		21	1		41	1		61	3		61	4	
2	4		22	4		42	2		62	3		62	3	
3	3		23	3		43	3		63	1		63	4	
4	2		24	3		44	3		64	4		64	1	
5	3		25	1		45	3		65	3		65	4	
6	1		26	2		46	2		66	3		66	3	
7	4		27	2		47	3		67	4		67	2	
8	2		28	2		48	1		68	3		68	3	
9	3		29	1		49	2		69	1		69	3	
10	4		30	2		50	3		70	1		70	2	
11	2		31	1		51	4		71	4		71	3	
12	1		32	1		52	3		72	4		72	1	
13	4		33	1		53	2		73	3		73	2	
14	1		34	4		54	2		74	4		74	1	
15	1		35	4		55	3		75	3		75	1	
16	4		36	4		56	4		76	3		76	3	
17	1		37	4		57	3		77	1		77	1	
18	1		38	1		58	4		78	4		78	2	
19	4		39	2		59	1		79	2		79	2	
20	2		40	3		60	3		80	2		80	1	

